

Estimation of the Three Phase Coexisting Line for Carbon Dioxide Type I Hydrates using Molecular Dynamics

José Manuel Míguez, María M. Conde and Jean-Philippe Torre

*Lab. des Fluides Complexes et Leurs Réservoirs, UMR 5150, Université de Pau et des Pays de l'Adour, Pau,
France*

Felipe J. Blas

Dpto de Física Aplicada, Univ. de Huelva, Huelva, Spain

Manuel M. Piñeiro^{C, S}

Dpto. de Física Aplicada, Univ. de Vigo, Vigo, Spain

mmpineiro@uvigo.es

Carlos Vega

Dpto. de Química Física I, Univ. Complutense, Madrid, Spain

The interest on CO₂ hydrates characterization has increased remarkably over the last few years due to the possibility of using them for waste CO₂ recovery, capture and storage. The feasibility of replacing CH₄ by CO₂ in hydrate deposits existing in nature has been discussed, and seems to be a promising alternative with evident environmental benefits in the near future. Nevertheless, the modelling of CO₂ or mixed hydrates has received less attention so far, and the theoretical estimation of their phase equilibria and thermophysical and transport properties still needs further contributions. In this case, rigid non polarizable molecular models have been used to describe H₂O and CO₂, and the three phase CO₂ Hydrate-Liquid H₂O-Liquid CO₂ equilibrium line has been determined using Molecular Dynamics and the direct coexistence technique. The influence of the initial occupancy rate of the different cells of the type I hydrate has been evaluated, as well as the effect of the different parameterizations of the molecular models in each case. It has been found that, although correct qualitative results are found for this type of models up to high pressures, the H₂O model version plays a key role in the precise determination of the phase boundary location, which is by far less sensitive to the CO₂ model version employed.