Acoustic Method as a Tool for Searching New Applications of Ionic Liquids

Małgorzata Musiał^{1, S, C} and Marzena Dzida²

¹Applied Chemicals and Materials Division, National Institute of Standard and Technology, Boulder, CO, U.S.A. ²Institute of Chemisty, University of Silesia in Katowice, Katowice, Poland malgorzata.musial@nist.gov

The design of industrial processes and the evaluation of applications for new products can only be achieved if their exact physicochemical properties are known. For the rational consideration of ionic liquids (ILs) as working fluids such as heat-transfer or hydraulic fluids on a larger or industrial scale, knowledge of the physicochemical properties of these fluids, both under ambient conditions and under high pressure, is necessary. [1,2] The highpressure data help develop technologies that require the exposure of working liquids to changing pressure. In that context, we have investigated the speed of sound, density, isobaric and isochoric heat capacities, isentropic and isothermal compressibility coefficients, and isobaric thermal expansion coefficient as a function of pressure and temperature. The $p\rho T$, pC_pT data, and derived properties were obtained using an acoustic method. This alternative to direct methods is regarded as a precise tool for investigating the thermodynamic properties of compressed liquids. [1] The ppT data obtained from the experimental speed of sound are considered to be the most reliable because the speed of sound can be measured accurately over a wide range of temperature and pressure values.

In this talk, the usefulness of ILs as heat-transfer media and hydraulic fluids will be analyzed and discussed in comparison with commercial working fluids. The discussion will include the above mentioned properties along with cytotoxicity, surface tension, and wettability.

Other affiliations of Małgorzata Musiał: Institute of Chemistry, University of Silesia in Katowice, Katowice, Poland; and Department of Physics, University of Colorado, Boulder, CO, United States

References

- 1. M. Dzida, E. Zorębski, M. Zorębski, M. Żarska, M. Geppert-Rybczyńska, M. Chorążewski, J. Jacquemin, I. Cibulka, Chem. Rev. 2017, 117, 3883.
- 2. M. Kambic, R. Kalb, T. Tasner, D. Lovrec, Sci. World J. 2014, 2014, 504762.
- 3.
- 4.
- 5.
- *6*.
- 7.
- 8.
- 9.
- 10.
- 11.
- 12.
- 13.

14. 15.

16.